## IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Appl. No. : 10/560,497 Confirmation No. 4578

Applicants: : Lars Terje Holmaas et ol. Filed : December 12, 2005

TC/A.U. : 1621

Examiner: : Sudhokar Katakam

Docket No. : PN0324 Customer No. : 36335

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

## **RULE 132 DECLARATION**

in occordance with rule 37 CFR 1.112, I, Ole Magne Homestad, do hereby declore that:

I am inventor of said invention and currently reside in Sør-Audnedal, Norway.

Reference is made to the Office Actions dated 11/21/2006, 04/26/2007, 10/29/2007 ond 04/14/2008.

In the present N-alkylation step of the synthesis of lohexal 2-methoxyethanol (2-ME) is used as solvent. When we searched for an alternative solvent to 2-ME, we considered the "nearest neighbour" of 2-ME, namely 2-ethoxyethanol. It was impossible to use this solvent due to solubility and selectivity. 1-methoksy-2-propanal (PM) was not considered since it was estimated to be even less point than 2-ethoxyethanol.

To be able to use a solvent in the alkylation step it is necessory that NaOH is soluble. This is the case for 2-ME. Regarding PM, NaOH was not soluble and hence, it would be impossible to dissolve '\$41' to obtain alkylation reaction conditions.

At first a modification of PM by adding co-solvents for increasing solubility of NaOH was not considered due to the assumption that such an addition would worsen selectivity of the

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reaction. This is true for 2-ME. It was therefore definitely not obvious to pursue this lead any further.

In summary it was not believed that PM was able to substitute 2-ME. This assumption was based on decades of experience with these processes and general chemical knowledge. It was consequently much unexpected that PM turned out to be a suitable replacement for 2-ME.

I declare under penalty of perjury pursuant to the laws of the United States of America that the foregoing is true and correct, and that this declaration was executed by me on 11th July 2008 at Spangereid, Norway.

Ole Magne Hamestad